Chemistry 211 Name: Chapter 16 Quiz #1 1. The name of the following ether is: (CH₃)₂CH-O-CH₂CH₂CH₂CH₃ 1) butyl isopropyl ether. butyl isopropyl ether.
 isobutyl propyl ether.
 butyl propyl ether.
 butyl propyl ether. 2) isobutyl propyl ether. 2. The C-O-C bond angle in dimethyl ether is closest to: 2) 109⁰ 3) 120⁰ 4) 180⁰ 1) 90⁰ 3. Match the boiling points with the following three isomers of $C_4H_{10}O_2$. CH₃OCH₂CH₂OCH₃ CH₃CH₂OCH₂CH₂OH HOCH₂CH₂CH₂CH₂CH₂OH 85⁰C 230⁰C 135⁰C 1) 85⁰C 135⁰C 2) 230⁰C 85⁰C 135⁰C 230⁰C 3) 4) 135⁰C 230⁰C 85⁰C 4. Of the following, which yields isopropyl methyl ether as the major product with little or no byproducts? 1) $(CH_3)_2 CHO^{-}Na^{+} + CH_3I \longrightarrow$ 2) $CH_3O^{-}Na^{+}$ + $(CH_3)_2CHI$ \longrightarrow 3) (CH₃)₂CHOH + CH₃OH $\xrightarrow{\text{H}_2\text{SO}_4}$ 4) all three give isopropy methyl ether as the major product 2) 2 3) 3 1) 1 4) 4 5. Which of the following is not a good method to make tert-butyl methyl ether? 1) $(CH_3)_3CO^{-}Na^{+} + CH_3Br \longrightarrow$ 2) $(CH_3)_3CBr + CH_3O^{-}Na^{+} \longrightarrow$ 3) $H_2C=C(CH_3)_2 + CH_3OH \longrightarrow$ 4) $(CH_3)_3CBr$ + CH_3OH heat 1) 1 2) 2 3) 3 4) 4

6. Which of the following is \underline{not} an intermediate in the reaction below?

$$2 \operatorname{CH_{3}OH} \xrightarrow{\operatorname{H_{2}SO_{4}, \text{heat}}} \operatorname{CH_{3}OCH_{3} + H_{2}O}$$

$$1) \operatorname{H_{3}C} \xrightarrow{\stackrel{\stackrel{\rightarrow}{O}-H}}_{H} 3) \operatorname{H} \xrightarrow{\stackrel{\stackrel{+}{C}-H}}_{H}$$

$$2) \operatorname{H_{3}C} \xrightarrow{\stackrel{\stackrel{\rightarrow}{O}-CH_{3}}}_{H} 4) \text{ they are all intermediates}$$

$$1) 1 2) 2 3) 3 4) 4$$

7. What are the products of the reaction below?



8. What is the product of the following reaction?



9. The reaction shown below can be described as an:



- 1) acid-base reaction followed by an intramolecular Williamson ether synthesis.
- 2) acid-base reaction followed by an intramolecular $\ensuremath{S_{\mathrm{N}}}\xspace1$ reaction.
- 3) E2 reaction followed by an addition reaction to a double bond.
- 4) $S_{\rm N}2$ reaction followed by an intramolecular Williamson ether synthesis.
- 10. What is the product of the following reaction?



- 1) (S)-1,2-propanediol
- 2) (R)-1,2-propanediol
- 3) racemic mixture of 1,2-propanediol
- 4) 1,3-propanediol
- 11. Which of the following reacts the fastest with NaOH, H_2O ?
 - 1) ethylene oxide (oxirane) 2) cis-2,3-dimethyloxirane
 - 3) trans-2,3-dimethyloxirane 4) 2,2,3,3-tetramethyloxirane

- 12. What reagents and/or reaction sequence below would convert *trans*-3-hexene to *meso*-3,4-hexanediol?
 - 1) OsO4, (CH₃)₃COOH, (CH₃)₃COH, NaOH
 - 2) $B_2H_6/diglyme$ followed by $H_2O_2/NaOH$
 - 3) O_3 followed by Zn/H_2O
 - 4) CH₃CO₃H followed by NaOH/H₂O
- 13. Which of the following yields an epoxide on treatment with NaOH?
 - 1) *cis*-2-bromocyclohexanol
 - 2) *trans-2-bromocyclohexanol*
 - 3) cis-1,2-cyclohexanediol
 - 4) 3-bromocyclohexene
- 14. Which of the following epoxides is formed when KOH is added to the optically active halohydrin shown below?



- 1) trans-(2S, 3S)-2, 3-dimethyloxirane
- 2) *trans*-(2R, 3R)-2, 3-dimethyloxirane
- 3) 2,2-dimethyloxirane
- 4) *meso-2*, 3-dimethyloxirane
- 15. Benzene reacts with 2-methyloxirane in the presence of $AlCl_3$ to give a product with a formula of $C_9H_{12}O$. Identify the product.





16. Propose a mechanism for the reaction shown below. (Hint: $\mbox{CF}_3\mbox{CO}_2\mbox{H}$ is a strong acid.)



Propose a synthesis of the bicycloalkene starting material above. (Hint: Furan can be used as a diene in Diels-Alder reactions. What dieneophile would work best?)

No.	in	No. on		
Q-B	ank	Test	orrect Answer	
16	1	1		
16	3	2		
16	5	3		
16	7	4		
16	9	5		
16	11	6		
16	13	7		
16	15	8		
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16	19	10		
16	21	11		
16	23	12		
16	25	13		
16	27	14		
16	29	15		
16	31	16		

Answer Key for Test "211c16q1.tst", 2/23/2004