1. What is the IUPAC name of the following compound? 1) 3-methyl-5-heptanone 2) 5-ethyl-3-hexanone 3) 5-methyl-3-heptanone 4) 2-ethyl-4-hexanone 2. Which of the following is an acceptable IUPAC name for the compound below? CHO Br 2) 6-bromo-3-chlorobenzaldehyde 1) *o*-bromo-*m*-chlorobenzaldehyde 1)o-bromo-m-chlorobenzaldehyde2)6-bromo-3-chlorobenzaldehyde3)2-bromo-5-chlorobenzaldehyde4)1-bromo-4-chlorobenzaldehyde 3. The carbon-oxygen bond of an aldehyde is formed by overlap of which two orbitals? 2) sp^2-sp^2 3) sp^2-2p 4) 2p-2p1) sp-sp 4. Identify the reagents needed to carry out the following conversion. $HC \equiv CCH_2CH_2CH_3 \xrightarrow{?} CH_3CCH_2CH_2CH_3$ 1) $H_2/Lindlar Pd$ followed by H_2SO_4/H_2O 2) O_3 followed by H_2O 3) H_2O , $HgSO_4/H_2SO_4$ 4) LiAIH₄ followed by H_2O 5. Which of the following reagents would carry out the isotopic substitution reaction shown below? $C \xrightarrow{16} CH_3 \xrightarrow{?} H_3C \xrightarrow{18} H_3C$ 1) ${}^{18}\text{O}_2/\text{Ni}$ (cat.) 2) $\text{H}_2{}^{18}\text{O}/\text{HCl}$ (cat.) 3) $\text{Cr}{}^{18}\text{O}_3/\text{pyridine}$ 4) ${}^{18}\text{O}_3$ 1) 1 2) 2 3) 3 4) 4

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6. What is the product of the reaction below?



7. What is the product of the reaction of butanal with excess methanol and catalytic sulfuric acid?

2) CH₃CH₂CH₂COCH₃

0

O || 1) CH₃CH₂CHCH | OCH3

 $3) CH_3CH_2CH_2CH_2OCH_3 \qquad 4) CH_3CH_2CH_2CH(OCH_3)_2$

1) 1 2) 2 3) 3 4) 4

8. Which one of the following is not an intermediate in the acid-catalyzed reaction of benzaldehyde with 2 equivalents of methanol to give benzaldehyde dimethyl acetal?



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9. What is the product of the reaction shown?



10. What are the products of the following reaction?



- 1) cyclohexanone and ethanol
- 2) cyclohexanone and ethanal
- 3) 1,2-cyclohexanediol and ethanal
- 4) 1,2-cyclohexanediol and ethanol
- 11. Which one of the following gives ethanal, $CH_3CH=O$, (as one of two products) when added to an aqueous solution of HCl?



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12. What is the product of the reaction below?

 $\underbrace{\bigcirc}_{O} \qquad \underbrace{(C_6H_5)_3 \overset{\dagger}{P} - \ddot{CH}_2}_{DMSO} \xrightarrow{\bullet}$

- 1) 2-methyl-1-pentene
- 2) 2-methyl-2-propyloxirane
- 3) 4-methyl-1-pentene
- 4) 1-pentene
- 13. Which of the following reacts with $(CH_3CH_2)_2NH$ to give the compound shown below?



14. Baeyer-Villiger oxidation reactions can use peroxycarboxylic acids to convert ketones to:

1)	carboxylic acids.	2)	esters.
3)	epoxides.	4)	-hydroxy ketones

15. What is the product of the following Baeyer-Villiger oxidation reaction?



16. Which of the following reacts with methylamine at the fastest rate?

1) 1-pentene 2) pentanal 3) 2-pentanone 4) 3-pentanone

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17. What is the product of the reaction sequence below?

$$\begin{array}{c|c} 1) \text{ BH}_3/\text{THF} & \text{PCC} & (C_6H_5)_3\text{P=CH}_2 \\ \hline 2) \text{ H}_2\text{O}_2, \text{ OH}^- & \text{CH}_2\text{Cl}_2 & \text{DMSO} \end{array}$$

1)2-methyl-1-hexene2)2,3-dimethyl-2-pentene3)2-methyl-2-hexene4)3-methyl-1-hexene

18. Propose a mechanism which accounts for the formation of the cyclic compound below.

	$P(C_6H_5)_3 Br$
(C ₆ H ₅) ₃ P=CHCH ₂ CH ₂ CH ₂ Br	