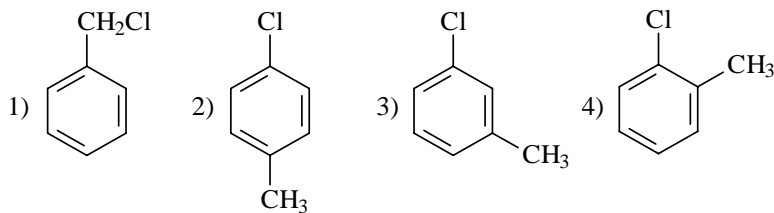


1. Which of the following has the weakest carbon-chlorine bond?

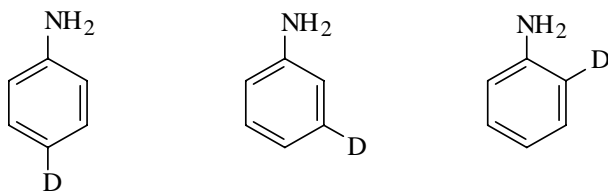
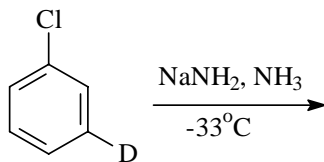


- 1) 1                      2) 2                      3) 3                      4) 4

2. Which one of the reagents readily reacts with bromobenzene?

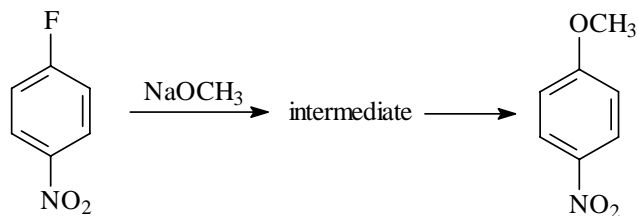
- 1)  $\text{NaOCH}_2\text{CH}_3$  at  $25^\circ\text{C}$                       2)  $\text{NaCN/DMSO}$  at  $25^\circ\text{C}$   
3)  $\text{NaNH}_2/\text{NH}_3$  at  $-33^\circ\text{C}$                       4)  $(\text{CH}_3)_2\text{NH}$  at  $25^\circ\text{C}$

3. Which of the following best estimates the percentages of the three isomeric deuterated anilines from the reaction shown below?



- |    |     |     |     |
|----|-----|-----|-----|
| 1) | 25% | 50% | 25% |
| 2) | 33% | 33% | 33% |
| 3) | 50% | 25% | 25% |
| 4) | 66% | 33% | 0%  |

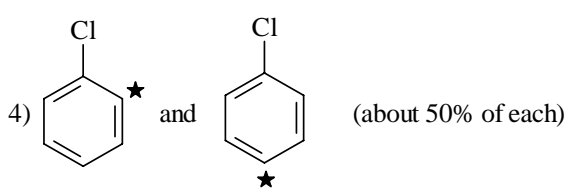
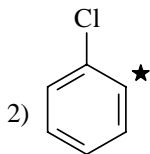
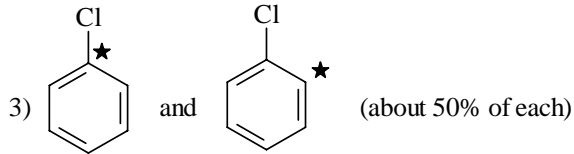
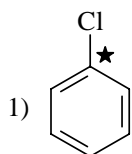
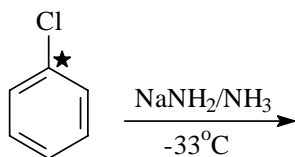
4. Which of the following is (are) true concerning the intermediate in the addition-elimination mechanism of the reaction below?



- A. The intermediate is aromatic.  
 B. The intermediate is a resonance stabilized anion.  
 C. Electron withdrawing groups on the benzene ring stabilize the intermediate.

1) only A                      2) only B                      3) A and C                      4) B and C

5. Carbon-14 labelled chlorobenzene is reacted with sodium amide in ammonia as shown below. Which of the following depicts the carbon-14 label in the product(s)?



1) 1                      2) 2                      3) 3                      4) 4

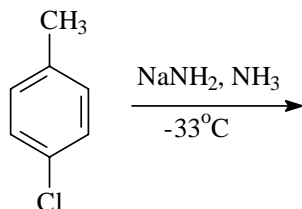
6. Which of the following reacts at the fastest rate with potassium methoxide (KOCH<sub>3</sub>) in methanol?

- 1) fluorobenzene                      2) 4-nitrofluorobenzene  
 3) 2,4-dinitrofluorobenzene                      4) 2,4,6-trinitrofluorobenzene

7. Which of the following is the kinetic rate equation for the addition-elimination mechanism of nucleophilic aromatic substitution?

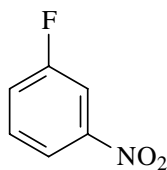
- 1) rate =  $k[\text{aryl halide}]$
- 2) rate =  $k[\text{nucleophile}]$
- 3) rate =  $k[\text{aryl halide}][\text{nucleophile}]$
- 4) rate =  $k[\text{aryl halide}][\text{nucleophile}]^2$

8. Which of the following best estimates the percentages of the three isomeric methylanilines from the reaction shown below?

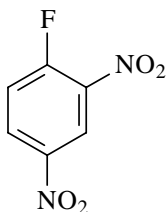


	<u>ortho</u> -methylaniline	<u>meta</u> -methylaniline	<u>para</u> -methylaniline
1)	33%	33%	33%
2)	40%	40%	20%
3)	0%	50%	50%
4)	0%	66%	33%

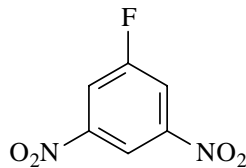
9. Arrange the following compounds in order of increasing reactivity with sodium methoxide,  $\text{NaOCH}_3$ ?



A



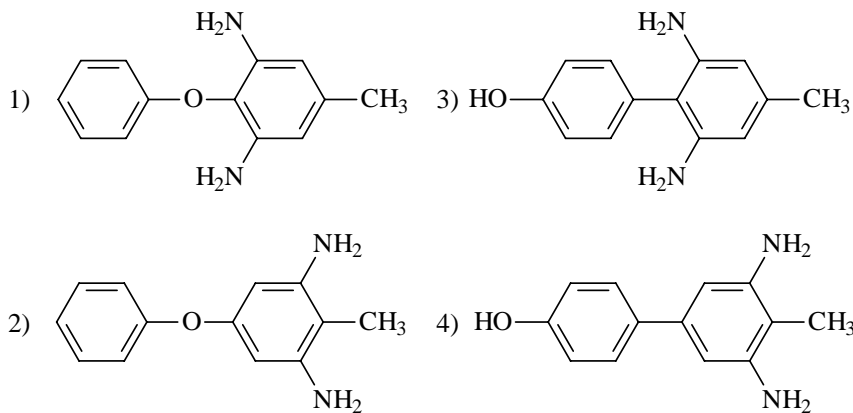
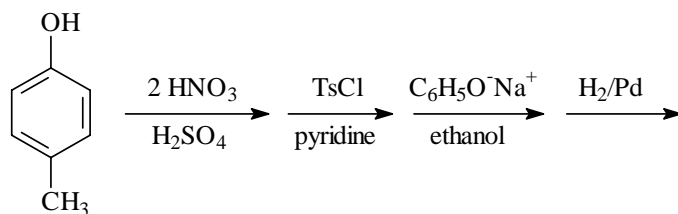
B



C

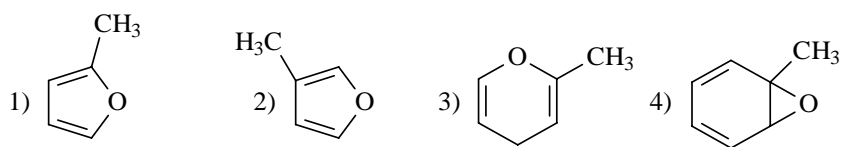
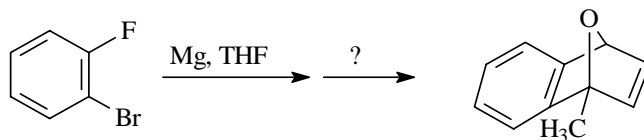
- 1)  $A < B < C$
- 2)  $A < C < B$
- 3)  $B < A < C$
- 4)  $C < B < A$

10. Which of the following is the product from the reaction sequence shown below?



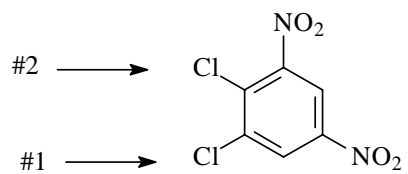
1) 1                      2) 2                      3) 3                      4) 4

11. Identify the diene required for the synthesis shown below.



1) 1                      2) 2                      3) 3                      4) 4

12. Which chlorine is most susceptible to nucleophilic substitution with  $\text{NaOCH}_3$  in methanol?



- 1) #1
- 2) #2
- 3) #1 and #2 are equally susceptible
- 4) no substitution is possible